Spin Crossover Complex Film, [Fe^{II}(H-trz)₃]-Nafion, with a Spin Transition around Room Temperature

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We have synthesized $[Fe^{II}(H-trz)_3]_n$ complex film by using ion-exchange resin(Nafion) as counter anion, and have found the spin crossover phenomenon about at 260 K. Moreover, we have proved the existence of straight one-dimensional Fe chain structure of $[Fe^{II}(H-trz)_3]_n$ on Nafion by means of EXAFS.

Recently, spin crossover phenomena converted between high spin (HS) state and low spin (LS) state have gained renewed importance since the discovery of the photo-induced spin transition called LIESST (Light Induced Excited Spin State Trapping) for [Fe^{II}(ptz)₆](BF)₂ (ptz = 1-propyltetrazole)¹ and the thermally induced spin transition with a large thermal hysteresis around room temperature for 1,2,4-triazole (= trz) bridged iron(II) complexes, [Fe^{II}(H-trz)_{2.85}(NH₂-trz)_{0.15}] (ClO₄)₂·H₂O.² These properties have attracted much attention from the viewpoint of the photonic molecular device.

From the viewpoint of molecular devices, it is important to control the spin transition temperature T_c and the hysteresis width ΔT . In the case of triazole bridged polymeric Fe^{II} complex system, $[Fe^{II}(R-trz)_3]X_2$, the control of T_c and ΔT has extensively been carried out. In this system, T_c and ΔT remarkably depend on not only ligand molecule³ (R-trz) but also counter anion⁴ and crystal water.⁵ In order to realize photonic devices, the synthesis of single crystal or transparent film is indispensable. However, single crystals of $[Fe^{II}(R-trz)_3]X_2$ have not yet been obtained. So that, we have tried to synthesize the spin crossover complex film based on [Fe^{II}(R-trz)₃]. The challenge to synthesize the spincrossover complex film has been done for $[Fe(2-mephen)_3]^{2+}$ (2mephen = 2-methyl-1,10-phenanthroline).⁶ However, our work is the first time to synthesize $[Fe^{II}(R-trz)_3]$ complex film with a spin transition around room temperature. Here, we report the synthesis and the spin crossover phenomenon for [Fe^{II}(R-trz)₃]-Nation ($R = H, NH_2$).

In the triazole-bridged iron(II) chain system, various kinds of molecules having R-SO₃⁻ ($R = C_n H_{2n+1}$, C₆H₅, etc.) are acceptable as counter anion, from which the idea to synthesize the spin crossover complex film by using Nafion film (DuPont Co.) has been raised. Nafion is composed of a polytetrafluor-oethylene backbone with perfluorinated ether side chains terminated by sulfonic acid group as shown in Figure 1. In the case of Nafion 117, the pendant Y-SO₃H (Y = perfluorinated ether side

Figure 1. Molecular formula of Nafion 117: x = 6, y = 1, z = 1. chain) groups in the perfluorinated ionomer are known to form clusters of 40 Å diameter separated by a distance of 50 Å and interconnected through channels of 10 Å when they are swollen in water.⁷

The spin crossover complex film, $[Fe^{II}(R-trz)_3]$ -Nafion (R = H, NH₂), was prepared in the following way. The acid form of Nafion was immersed in an aqueous solution of FeSO₄. The mobile anions (SO₄²⁻) are more or less completely excluded from the polymer matrix. Only Fe²⁺ ions are absorbed into Nafion membrane. After being immersed in the solution for 2 h, the membrane was rinsed in methanol, and then it was immersed in methanol solution of R-trz at 330 K for 1 h. All operations were carried out under nitrogen gas. The preparation process of $[Fe^{II}(R-trz)_3]$ -Nafion is schematically shown in Figure 2.



Figure 2. Schematic preparation process of [Fe^{II}(R-trz)₃]-Nafion. N-N represents 4-R-trz.

The colors of Nafion films after being immersed in methanol solution of NH₂-trz and H-trz are colorless and pale pink, respectively, at 290 K. However, their color became to be purple at liquid N₂ temperature, which implies the spin crossover phenomenon (Inset in Figure 3). Note that the colors of the HS state $(t_{2g}^4 e_g^2, S = 2)$ and the LS state $(t_{2g}^6, S = 0)$ of $[Fe^{II}(R-trz)_3]X_2 \cdot nH_2O$ (R = H, NH₂; X = CIO₄, etc.) are colorless and purple, respectively. In fact, the absorption spectrum of $[Fe^{II}(H-trz)_3]$ -Nafion shows two broad bands at 18500 cm⁻¹ and 24500 cm⁻¹ corresponding to the ${}^{1}A_{1g} \rightarrow {}^{1}T_{1g}$ and ${}^{1}A_{1g} \rightarrow {}^{1}T_{2g}$ transitions, respectively, in the LS state (S = 0), while it shows a broad band at 12500 cm⁻¹ corresponding to the ${}^{5}T_{2g} \rightarrow {}^{5}E_{g}$ transition in the HS state (S = 2).

In order to prove that the one-dimensional Fe chain structure of $[Fe^{II}(H-trz)_3]_n$ is formed on Nafion film, we measured the Fe *K*edge EXAFS spectra for $[Fe^{II}(R-trz)_3]$ -Nafion (R = H, NH₂). The Fe K-edge EXAFS spectra were taken in the conventional transmission mode at Beam Line-10B in the Photon Factory in the Institute of Materials Structure Science, Tsukuba. The Fourier transforms of each spectrum at 65 K are shown in Figure 4. The main peak at about 1.7 Å of the Fourier transform corresponds to the Fe-N scattering. A noticeable feature is the existence of peak around 7 Å. In the normal EXAFS spectroscopy, any distinct peak does not appear in this range. However, in the case that the Fe atoms are arranged in a straight chain, the multiple scattering is

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Figure 3. χT as a function of temperature for [Fe^{II}(H-trz)₃]-Nafion, [Fe^{II}(NH₂-trz)₃]-Nafion in cooling mode ($\mathbf{\nabla}$, ∇) and heating mode ($\mathbf{\Delta}$, Δ), respectively. The susceptibility of the vertical axis is the value per unit mass which includes the membrane mass. Inset shows the color (purple) in the low-spin state (T = 77 K) and that (colorless) in the high-spin state (T = 300 K).



Figure 4. Fourier transforms of Fe *K*-edge EXAFS oscillation function $k^3\chi(k)$ for (a) [Fe^{II}(H-trz)₃]-Nafion and (b) [Fe^{II}(NH₂-trz)₃]-Nafion at 65 K. Inset shows the multiple scattering from the next nearest neighbor Fe-Fe-Fe shell which is responsible for the EXAFS spectrum at about 7 Å.

generally enhanced, so that the significant peak appears at long radius region.⁸ The peak at about 7 Å corresponds to the Fe-Fe-Fe multiple scattering, which proves the existence of straight one-dimensional Fe chain structure.

In order to investigate the detailed spin crossover phenomenon, we measured the temperature dependence of magnetic susceptibility(χ) for [Fe^{II}(H-trz)₃]-Nafion film and [Fe^{II}(NH₂trz)₃]-Nafion film. The static magnetic susceptibility was measured by using a Quantum Design MPMSXL SQUID susceptometer. The sample was sealed in an aluminum capsule to avoid a loss of crystal water in heating process. The applied magnetic field was 1 kG. The temperature was swept in quite slow rate of 0.1 K/min for heating and cooling process in order to investigate the thermal hysteresis at the spin transition. The χT as a function of temperature for [Fe^{II}(H-trz)₃]-Nafion film and [Fe^{II}(NH₂-trz)₃]-Nafion are shown in Figure 3. The spin transition for [Fe^{II}(H-trz)₃]-Nafion takes place at about $T_{1/2} = 260$ K ($T_{1/2}$ is defined as the temperature at which the fraction of HS induced by the spin transition is a half.). The thermal hysteresis width is confirmed to be 3 K. The spin transition for [Fe^{II}(NH₂-trz)₃]-Nafion takes place at about $T_{1/2} = 198$ K. In this case, the thermal hysteresis vanishes.

The magnetic susceptibility as a function of temperature shows residual paramagnetic fraction below 200 K, which is attributed to the HS state of Fe^{II}. The line profile of ESR below 10 K is typical one of the HS state of Fe^{II}. The residual HS fraction would be attributed to the terminal Fe^{II} site in the oligomer of [Fe(H-trz)₃]_n on Nafion film. In connection with this, the following should be mentioned. In the case of the Fe^{II} trimer complex, [Fe^{II}₃(Et-trz)₆(H₂O)₆](CF₃SO₃)₆ (Et-trz = 4-ethyl-1,2,4-triazole), the central Fe^{II} site undergoes the LS-HS transition at about 200 K, while the spin state of the terminal Fe^{II} sites is the HS state between 2 K and 300 K.⁹

In conclusion, we have succeeded in synthesizing transparent $[Fe^{II}(R-trz)_3]$ complex film by using ion-exchange film (Nafion) as counter anion, which shows the spin crossover phenomenon at about 260 K. In our preparation process, it is possible to prepare the homogeneous $[Fe^{II}(R-trz)_3]$ complex film of 300 mm × 300 mm. The development of transparent $[Fe^{II}(R-trz)_3]$ -Nafion film will open a large field of photonic molecular devices based on spin-crossover phenomenon.

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References

- 1 S. Decurtins, P. Gütlich, C. P. Köhler, H. Spiering, and A. Hauser, *Chem. Phys. Lett.*, **105**, 1 (1984).
- 2 J. Kröber, E. Codjovi, O. Kahn, F. Grolière, and C. Jay, J. Am. Chem. Soc., 115, 9810 (1993).
- 3 O. Kahn and C. J. Martinez, *Science*, **279**, 44 (1998).
- 4 J. J. A. Kolnaar, Doctoral Thesis, Leiden University (1998).
- 5 S. Toyazaki, Y. Murakami, T. Komatsu, and N. Kojima, *Mol. Cryst. Liq. Cryst.*, **343**, 175 (2000).
- 6 A. Hauser, J. Adler, and P. Gütlich, *Chem. Phys. Lett.*, **152**, 468 (1988).
- 7 T. D. Gierke, G. E. Munn, and F. C. Wilson, J. Polym. Sci., 19, 1687 (1981).
- 8 A. Stern, in "X-ray Absorption," ed. by D. C. Koning Sberger and R. Prins, John Wiley & Sons, New York (1988), Chap. 1.
- 9 G. Vos, R. A. G. De Graaff, J. G. Haasnoot, A. M. Van der Kraan, P. De Vaal, and J. Reedijk, *Inorg. Chem.*, 23, 2905 (1984).